ERRATA SHEET FOR ANSI/ASHRAE STANDARD 34-2019
Designation and Safety Classification of Refrigerants

February 8, 2024

The corrections listed in this errata sheet apply to ANSI/ASHRAE Standard 34-2019. The first printing is identified on the outside back cover as "Product code: 86015 5/19". Shaded items have been added since the previous published errata sheet date August 4, 2022 was distributed.

<u>**Page**</u>	<u>**Erratum**</u>
30	**B2.4.1 Leaks Under Storage/Shipping Conditions.** In Section B2.4.1 change the equation in the third sentence, $T = T_b + 0.8(T_b - T_c)$, to $T = T_b + 0.8(T_c - T_b)$ as shown below. Changes are highlighted in yellow. *(Note: Additions are shown in <u>underline</u> and deletions are shown in ~~strikethrough~~.)* **B2.4.1 Leaks Under Storage/Shipping Conditions.** To simulate leaks under storage/shipping conditions, the container shall be filled with the WCF to 90%, by mass, of the maximum fill. The maximum fill for fluids having a critical temperature greater than 130°F (54.4°C) is the calculated mass that gives a 100% liquid fill at 130°F (54.4°C). The maximum fill for fluids whose critical temperature is lower than 130°F (54.4°C) is the calculated mass that gives 100% liquid fill at temperature <u>$T = T_b + 0.8(T_c - T_b)$</u> ~~$T = T_b + 0.8(T_b - T_c)$~~, where T_b is the bubble-point temperature at atmospheric pressure (101.3 kPa) and T_c is the fluid critical temperature. The charged blend shall be vapor leaked, 2% by mass of the initial charge per hour, at the following temperatures:
35	**Table D-2.** Change Composition (Mass %) for refrigerant number 421A from "R-125/134a (58.0/45.0)" to "R-125/134a (58.0/42.0)".
40	**Table E-1, Toxicity Table for Standard 34—ATEL, ODL, FCL, and RCL Values for Single-Compound Refrigerants (ppm v/v).** Change the Anesthesia NOEL value for 152a from 100,000 ppm to 66,400 ppm.
43	**F2. Heat of Combustion for a Refrigerant Blend.** In Equation F-7 relocate the floating minus sign "-" so that it precedes "{[2(0.2311) + 3(0.7689)] [–393.51]" as shown in the corrected equation below.

$$\Delta h_{combustion} =$$

$$\sum \Delta h_f(\text{reactants}) - \sum \Delta h_f(\text{products}) =$$

$$\{\Delta h_f(C_{2x+3y}H_{x+8y}F_{5x}) + (x+5y)\Delta h_f(O_2)\}$$

$$-\{(2x+3y)\Delta h_f(CO_2) + (5x)\Delta h_f(\text{HF})$$

$$+ [(x+8y-5x)/2]\Delta h_f(H_2O)\} =$$

$$-335.77 + [0.2311 + 5(0.7689)][O]$$

$$-\{[2(0.2311) + 3(0.7689)][-393.51]$$

$$+ [5(0.2311)][-273.30]$$

$$+ [0.5] [0.2311 + 8(0.7689)$$

$$-5(0.2311)[-241.83]\}$$

$$= 1701.6 \text{ kJ/mol}$$

(F-7)